

This article was downloaded by:

On: 27 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



## Organic Preparations and Procedures International

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t902189982>

### A RAPID AND EFFICIENT SYNTHESIS OF 2H-1,4-BENZOXAZINE-3(4H)-THIONES AND 2H-1,4-BENZOTHIAZINE-3(4H)-THIONES

D. R. Shridhar<sup>a</sup>; M. Jogibhukta<sup>a</sup>; V. S. H. Krishnan<sup>a</sup>

<sup>a</sup> IDPL Research Centre, Indian Drugs and Pharmaceuticals Limited, Hyderabad, INDIA

**To cite this Article** Shridhar, D. R. , Jogibhukta, M. and Krishnan, V. S. H.(1984) 'A RAPID AND EFFICIENT SYNTHESIS OF 2H-1,4-BENZOXAZINE-3(4H)-THIONES AND 2H-1,4-BENZOTHIAZINE-3(4H)-THIONES', *Organic Preparations and Procedures International*, 16: 2, 91 – 96

**To link to this Article:** DOI: 10.1080/00304948409356171

**URL:** <http://dx.doi.org/10.1080/00304948409356171>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

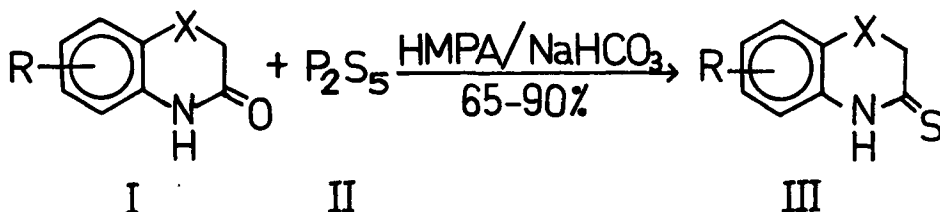
The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

A RAPID AND EFFICIENT SYNTHESIS OF 2H-1,4-BENZOXAZINE-3(4H)-  
THIONES AND 2H-1,4-BENZOTHAIAZINE-3(4H)-THIONES<sup>+</sup>

D. R. Shridhar<sup>\*</sup>, M. Jogibhukta and V. S. H. Krishnan

IDPL Research Centre  
Indian Drugs and Pharmaceuticals Limited  
Hyderabad-500 037, INDIA

Some 2H-1,4-benzoxazine- and benzothiazine-3(4H)-  
thiones were needed in substantial quantities for the prepa-  
ration of imidazo(2,1-c)(1,4)benzoxazine and benzothiazine  
derivatives of medicinal interest. Previous syntheses<sup>1-5</sup> of  
some of these thiolactams (IIIa-j) involving the treatment of  
the corresponding lactams (Ia-j) with P<sub>2</sub>S<sub>5</sub> in solvents such as  
xylene, pyridine, acetonitrile, CH<sub>2</sub>Cl<sub>2</sub> etc. suffer from dis-  
advantages such as long reaction times (24-72 hrs), formation  
of difficultly purifiable mixtures of products and variable and



often poor yields. Although Lawesson's reagent<sup>6,7</sup> was found  
to be generally satisfactory for the thionation of lactams  
like Ia-j which do not carry an additional carbamoyl function,  
neither this method nor the other thionation procedures<sup>1-5</sup>  
were found to be suitable for the selective thionation of the

lactams (Ik-n) to give the desired thiolactams (IIIk-n). In most of these cases, the required products were found to be generally contaminated with substantial quantities of the unwanted dithio derivatives (IIIk & l, R = NHCSCH<sub>3</sub>; IIIm & n, R = NHCSC<sub>6</sub>H<sub>5</sub>, X = O) resulting from the concomitant thionation of the other carbamoyl group which required tedious purification procedures for the isolation of the desired products. Recently Smolders *et al.*<sup>8</sup> reported the preparation of some 9-acridonethiones by treatment of the corresponding 9-oxo analogues with P<sub>2</sub>S<sub>5</sub> in hexamethylphosphoric triamide (HMPA). When applied to the preparation of some of the title compounds, this method gave the desired products of relatively low purity in 50-60% yields.

We now report that the use of NaHCO<sub>3</sub> as a basic additive in Smolders' procedure results in both rapid and selective thionation of the lactam carbonyl function to give the desired thiolactams (IIIa-n) in excellent yields (65-90%). The physical data of some of these thiolactams (IIIa-j) are reported in the Table.

A mixture of P<sub>2</sub>S<sub>5</sub> and NaHCO<sub>3</sub> readily dissolves in HMPA with the evolution of CO<sub>2</sub>. The resulting solubilized and presumably more reactive thionation agent could be expected to attack preferentially the comparatively more reactive<sup>9</sup> lactam carbonyl group of Ik-n thus perhaps explaining the facile and selective thionation.

The mild reaction conditions (heating the components in HMPA at 80° for 0.5-1 hr), the easy work-up, the selective thionation of only the lactam carbonyl function in lactams

carrying other carbamoyl groups, the high yields and purity of the products make the present synthesis a rapid and efficient general method for the preparation of a variety of benzoxazine and benzothiazine-3-thiones.

TABLE. Physical Data of Thiolactams IIIa-j

Compd. No.	X	R	Yield <sup>a</sup> (%)	mp. <sup>b</sup> (°C)	lit. mp. (°C)
IIIa	O	H	90	120	119-120 <sup>6</sup>
IIIb	O	6-Cl	85	195-196	195-196 <sup>6</sup>
IIIc	O	6-Me	87	204-206	204-206 <sup>5</sup>
IIId	O	6-NO <sub>2</sub>	80	204-206	204-206 <sup>5</sup>
IIIe	O	7-NO <sub>2</sub>	82	202-204	201-203 <sup>10</sup>
IIIf	S	H	90	128	128-129 <sup>6</sup>
IIIg	S	7-Cl	88	206-208	206-208 <sup>6</sup>
IIIh	S	7-Br	86	207-209	206-208 <sup>5</sup>
IIIi	O	6-NHCOOCH <sub>3</sub>	80	202	202 <sup>5</sup>
IIIj	O	7-NHCO <sub>2</sub> CH <sub>3</sub>	78	248	248 <sup>5</sup>

a) Isolated yields starting from compounds I. TLC showed absence of any other products. b) mps. are determined by means of a capillary and are uncorrected.

#### EXPERIMENTAL SECTION

Melting points were determined in glass capillaries on a Gallenkamp melting point apparatus and are uncorrected. Elemental analyses were performed using a Hosli microcombustion apparatus MK 101. Mass spectra were recorded on a Varian MAT CH 7A mass spectrometer. IR spectra were taken on a Perkin-Elmer 577 spectrophotometer and the NMR spectra were recorded on a Varian A-90 (EM-390) spectrometer. The starting lactams I were prepared following the methods reported earlier.<sup>11-14</sup>

**CAUTION:** As HMPA is known to be a potent animal carcinogen and is also suspected to be a human carcinogen; contact with skin and exposure to its vapour must be scrupulously avoided and all operations must be performed in an efficient hood.

**General Procedure** .- The appropriate lactam (I, 0.1 mol) was

added to a solution of  $P_2S_5$  (II, 26.64 gm, 0.12 mol) and  $NaHCO_3$  (25.2 gm, 0.3 mol) in HMPA (100 ml) and the resulting solution was stirred at  $80^\circ$  for 0.5 to 1 hr. It was then cooled to room temperature and added to ice cold water (2000 ml) when the required product separated as a solid. It was filtered, washed with water and recrystallised from either aqueous ethanol (IIIa-j) or acetone/pet.ether (IIIk-n) to give the corresponding thiolactam III.

6-Acetamido-2H-1,4-benzoxazine-3(4H)-thione (IIIk, X = O, R = 6-NHCOCH<sub>3</sub>), 70% yield, mp.  $142-144^\circ$ ; IR (nujol): 3340 (NH),  $1670\text{ cm}^{-1}$  (CONH); NMR (DMSO-d<sub>6</sub>):  $\delta$  2.1 (s, 3H, CH<sub>3</sub>), 4.83 (s, 2H, CH<sub>2</sub>), 6.7-7.6 (m, 3H, ArH), 10.13 (bs, CONH, exchangeable with D<sub>2</sub>O), 12.8 (bh, 1H, CSNH, exchangeable with D<sub>2</sub>O); MS: m/e 222 (M<sup>+</sup>), 179 (M-CH<sub>3</sub>CO)<sup>+</sup>, 43 (CH<sub>3</sub>CO).

Anal. Calcd. for C<sub>10</sub>H<sub>10</sub>N<sub>2</sub>O<sub>2</sub>S : C, 54.05; H, 4.50; N, 12.64  
Found : C, 54.29; H, 4.66; N, 12.76

7-Acetamido-2H-1,4-benzoxazine-3(4H)-thione (IIIl, X = O, R = 7-NHCOCH<sub>3</sub>), 65% yield, mp.  $148-150^\circ$ ; IR (nujol): 3280 (NH),  $1675\text{ cm}^{-1}$  (CONH); NMR (DMSO-d<sub>6</sub>):  $\delta$  2.0 (s, 3H, CH<sub>3</sub>), 4.72 (s, 2H, CH<sub>2</sub>), 7.0-7.3 (m, 2H, ArH), 7.4 (s, 1H, ArH), 10.0 (bs, 1H, CONH, exchangeable with D<sub>2</sub>O), 12.6 (h, 1H, CSNH, exchangeable with D<sub>2</sub>O); MS: m/e 222 (M<sup>+</sup>), 179 (M-CH<sub>3</sub>CO)<sup>+</sup>, 43 (CH<sub>3</sub>CO).

Anal. Calcd. for C<sub>10</sub>H<sub>10</sub>N<sub>2</sub>O<sub>2</sub>S : C, 54.05; H, 4.50; N, 12.64  
Found : C, 53.93; H, 4.61; N, 12.52

6-Benzamido-2H-1,4-benzoxazine-3(4H)-thione (IIIIm, X = O, R = 6-NHCOC<sub>6</sub>H<sub>5</sub>), 75% yield, mp.  $223-225^\circ$ ; IR (nujol): 3300 (NH),

BENZOXAZINE 3(4H)-THIONES AND 2H-1,4-BENZOTHAZINE 3(4H)-THIONES

1650  $\text{cm}^{-1}$  (CONH); NMR (DMSO- $d_6$ ):  $\delta$  4.8 (s, 2H,  $\text{CH}_2$ ), 7.0-8.1 (m, 8H, ArH), 10.3 (bs, 1H, CONH, exchangeable with  $\text{D}_2\text{O}$ ), 12.8 (bs, 1H, CSNH, exchangeable with  $\text{D}_2\text{O}$ ); MS: m/e 284 ( $\text{M}^+$ ), 105 ( $\text{C}_6\text{H}_5\text{CO}$ ).

Anal. Calcd. for  $\text{C}_{15}\text{H}_{12}\text{N}_2\text{O}_2\text{S}$  : C, 63.30; H, 4.20; N, 9.80

Found : C, 63.52; H, 4.38; N, 9.62

7-Benzamido-2H-1,4-benzoxazine-3(4H)-thione (IIIIn, X = O, R = 7-NHCOC $_6\text{H}_5$ ), 72% yield, mp. 205°; IR (nujol): 3340 (NH), 1660  $\text{cm}^{-1}$  (CONH); NMR (DMSO- $d_6$ ):  $\delta$  4.72 (s, 2H,  $\text{CH}_2$ ), 7.1-8.2 (m, 8H, ArH), 10.25 (bs, 1H, CONH, exchangeable with  $\text{D}_2\text{O}$ ), 12.68 (h, 1H, CSNH, exchangeable with  $\text{D}_2\text{O}$ ); MS: m/e 284 ( $\text{M}^+$ ), 105 ( $\text{C}_6\text{H}_5\text{CO}$ ).

Anal. Calcd. for  $\text{C}_{15}\text{H}_{12}\text{N}_2\text{O}_2\text{S}$  : C, 63.30; H, 4.20; N, 9.80

Found : C, 63.15; H, 4.36; N, 10.02.

Acknowledgement.- The authors wish to thank the staff of the Instrumentation Division of this Centre for spectral data and elemental analyses.

REFERENCES

- + Communication No.74 from IDPL Research Centre, Hyderabad-500 037, India.
1. M. Mazharuddin and G. Thyagarajan, *Tetrahedron*, **25**, 517 (1969).
  2. M. Pesson, *Fr. M.* 7614 (19 Jan 1970); *Chem. Abstr.*, **77**, 19653b (1972).
  3. C. Someswara Rao, M. P. Dave, P. N. Mody and A. P. Pandya, *Ind. J. Chem.*, **14B**, 999 (1976).
  4. C. Someswara Rao and M. P. Dave, *J. Ind. Inst. Sci.*, **59**, 94 (1977); *Chem. Abstr.*, **88**, 89601j (1978).
  5. D. R. Shridhar, M. Jogibhukta and V. S. H. Krishnan, *Ind. J. Chem.*, **21B**, 130 (1982).

6. D. R. Shridhar, C. V. Reddy Sastry, L. C. Vishwakarma and G. K. A. S. S. Narayan, *Org. Prep. Proc. Int.*, 12, 203 (1980).
7. D. R. Shridhar, S. S. Gandhi, K. Srinivasa Rao, A. N. Singh, H. N. Tripathi, S. Kondaiah and G. S. T. Sai, *Ind. J. Chem.*, 22B, 303 (1983).
8. R. R. Smolders, J. Hanuise, R. Coomans, V. Proietto, N. Voglet and A. Waefelaer, *Synthesis*, 493 (1982).
9. T. Kato, A. Takada and T. Ueda, *Chem. Pharm. Bull. Japan*, 20, 901 (1972).
10. D. R. Shridhar, S. S. Gandhi and K. Srinivasa Rao, *Ind. J. Chem.*, 20B, 1075 (1981).
11. G. Newbery and M. A. Phillips, *J. Chem. Soc.*, 3049 (1928).
12. L. Conti and G. Leandri, *Boll. Sci. fac. chim. ind. Bologna*, 15, 33 (1957); *Chem. Abstr.*, 51, 17926c (1957).
13. A. S. Angeloni and G. Pappalardo, *Gazz. Chim. Ital.*, 91, 633 (1961); *Chem. Abstr.*, 56, 10136e (1962).
14. D. R. Shridhar, M. Jogibhukta and V. S. H. Krishnan, *Org. Prep. Proc. Int.*, 14, 195 (1982).

(Received December 14, 1983; in revised form February 24, 1984)